

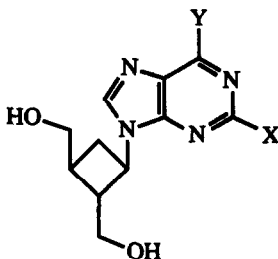
## Synthesis of [1 $\alpha$ ,2 $\beta$ ,3 $\alpha$ ]-5-amino-[2,3-bis(benzoyloxymethyl)cyclobutyl]imidazoles : important precursors to new anti-viral purine nucleosides.

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**Abstract :** Two new carbocyclic imidazoles, **6** and **9** have been synthesised *via* a simple method from the corresponding formamidine **5** ; these intermediates have been shown to be useful for the synthesis of new cyclobutyl nucleoside analogues including the known bis-benzoyl cyclobut-A **11**.

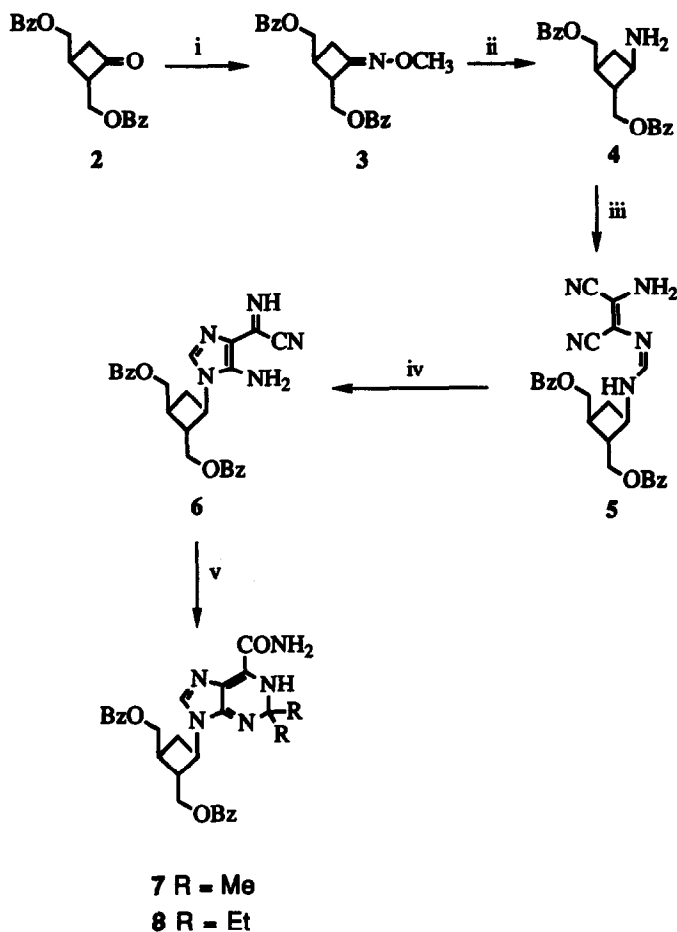
Recently, there has been much interest in the potent anti-viral agents cyclobut-A **1a** and cyclobut-G **1b**, carbocyclic analogues of the natural Oxetanocin<sup>1</sup>, and several syntheses have been reported.<sup>2,3</sup> Specific attention has been given to these compounds as they exhibit high activity against HIV infections.<sup>4</sup> However, all of these syntheses rely upon the coupling of a pre-formed purine ring to the carbocyclic moiety and this usually takes place only in moderate yield under forcing conditions. For some time we have been interested in the chemistry of N-substituted (2-amino-1,2-dicyanovinyl) formamidines as useful precursors to purine, imidazole and triazepine heterocycles<sup>5</sup> and we recognised that this approach could be used for a flexible synthesis of various purine derivatives of **1a**. Herein, we report the synthesis of two new imidazole intermediates, **6** and **9** and their use in the formation of new carbocyclic purine nucleosides.



**1a**; X = H, Y = NH<sub>2</sub>

**1b**; X = NH<sub>2</sub>, Y = OH

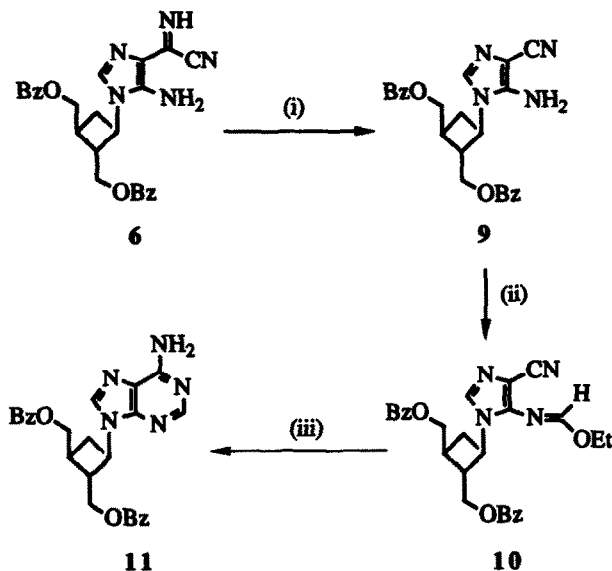
Racemic cyclobutanone **22** was treated with methoxylamine hydrochloride in pyridine to give the methyl oxime **3** as a mixture of geometrical isomers in 91% yield. Reduction of the oxime with sodium



**Scheme 1** Reagents and conditions: i, Methoxylamine hydrochloride, pyridine, r.t., 91% ; ii, sodium trifluoroacetoxy borohydride, THF, r.t., 43% ; iii, ethyl (Z)-[2-amino-1,2-dicyanovinyl]formimidate, EtOAc, Cat.  $\text{PhNH}_3^+\text{Cl}^-$ , r.t., 80% ; iv, 0.1 equiv. DBU,  $\text{CHCl}_3$ ,  $0^\circ\text{C}$ , 100% ; v, excess RCOR, r.t., R = Me, 92% ; R = Et, 88%.

trifluoroacetoxy borohydride<sup>6</sup> furnished the amine 4 in 43% yield after chromatography. Reaction of amine 4 with ethyl-(Z)-[2-amino-1,2-dicyanovinyl]formimidate<sup>5c</sup> in the presence of a catalytic amount of anilinium hydrochloride in ethyl acetate gave the amidine 5 as a foam in 80% yield.<sup>7</sup> Cyclisation of 5 with a catalytic amount of DBU in chloroform at  $0^\circ\text{C}$  gave the cyanoformimidoyl imidazole 6 in quantitative yield. This

compound is an important intermediate for the synthesis of a variety of purines.<sup>5a-c</sup> For example, **6** reacts with acetone to give the orange dihydropurine **7** in 92% yield and in a similar fashion treatment of **6** with pentan-3-one leads to **8** (88%). Treatment of **6** with an excess of DBU affords the amino-nitrile **9** (58%). This type of intermediate also represents an important purine precursor as by standard methodology<sup>8</sup> it can be transformed into a plethora of purine analogues. This is exemplified by a simple synthesis of the racemic form of the known bis-benzoyl derivative of cyclobut-A **11**<sup>9</sup> (Scheme 2). Reaction of **9** with triethylorthoformate gave the crude imidate **10**, which was cyclised by treatment with ammonia in anhydrous ethanol to give **11** in 65% overall yield from **9**.



**Scheme 2** Reagents and conditions: i, 2 equiv. DBU,  $\text{CHCl}_3$ , r.t., 20h, 58% ; ii,  $\text{CH}(\text{OEt})_3$ , 70-80°C ; iii,  $\text{NH}_3$ , EtOH, r.t., 65%.

Although the syntheses presented herein were performed with racemic materials, our methodology clearly demonstrates its use for easy access to derivatives of the potent anti-viral agents **1a** and **1b** for purposes of biological testing.

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7. All new compounds gave satisfactory spectroscopic data. Representative data for selected compounds: For 5:  $^1\text{H}$  NMR (300 MHz,  $(\text{CD}_3)_2\text{SO}$ )  $\delta$ : 1.86 (m, 1H), 2.39-2.60 (m, 3H), 4.40-4.55 (m, 5H), 6.12 (br.s, 2H), 7.55-7.69 (m, 5H), 7.70-8.00 (m, 2H), 7.98-8.14 (m, 4H), 8.24 (m, 1H). For 6:  $^1\text{H}$  NMR (300 MHz,  $(\text{CD}_3)_2\text{SO}$ )  $\delta$ : 2.29 (m, 1H), 2.57 (m, 1H), 2.77 (m, 1H), 3.22 (m, 1H), 4.40-4.70 (m, 5H), 6.85 (br.s, 2H), 7.52 (t,  $J = 7.5$  Hz, 2H), 7.61 (t,  $J = 7.5$  Hz, 2H), 7.67-7.78 (m, 3H), 7.92 (d,  $J = 7.5$  Hz, 2H), 8.09 (d,  $J = 7.5$  Hz, 2H), 11.0 (br.s, 1H). For 7:  $^1\text{H}$  NMR (300 MHz,  $(\text{CD}_3)_2\text{SO}$ )  $\delta$ : 1.37 (s, 3H), 1.42 (s, 3H), 2.30-2.60 (m, 3H), 3.21 (m, 1H), 4.38-4.62 (m, 5H), 6.29 (br. s, 1H), 7.48-7.80 (m, 7H), 7.85-8.15 (m, 5H), 8.28 (br. s, 1H) ppm. For 8:  $^1\text{H}$  NMR (300 MHz,  $(\text{CD}_3)_2\text{SO}$ )  $\delta$ : 0.81 (t,  $J = 7$  Hz, 3H), 0.88 (t,  $J = 7$  Hz, 3H), 1.48-1.76 (m, 4H), 2.40-2.50 (m, 3H), 3.28 (m, 2H), 4.38-4.61 (m, 5H), 5.88 (br.s, 1H), 7.48-8.11 (m, 12H), 8.25 (br.s, 1H). For 9:  $^1\text{H}$  NMR (300 MHz,  $(\text{CD}_3)_2\text{SO}$ )  $\delta$ : 2.18 (m, 1H), 2.55 (m, 1H), 2.74 (m, 1H), 3.20 (m, 1H), 4.40-4.70 (m, 5H), 6.32 (br.s, 2H), 7.60 (m, 4H), 7.74 (m, 3H), 7.95 (d,  $J = 7.7$  Hz, 2H), 8.07 (d,  $J = 7.5$  Hz, 2H).
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